

DOI: 10.1002/anie.200905997

## Ferroelectricity in Perovskites with s<sup>0</sup> A-Site Cations: Toward Near-Room-Temperature Multiferroics\*\*

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Magnetic ferroelectric (FE) materials (that is, multiferroics) have received much attention and can have wide technological applications owing to the possibility of controlling their electric properties and their magnetic properties by electric fields.<sup>[1,2]</sup> An important class of multiferroic and FE<sup>[3]</sup> materials is based on perovskites, ABO3, with two different cations A and B (for example A<sup>2+</sup> and B<sup>4+</sup> or A<sup>3+</sup> and B<sup>3+</sup>), in which corner-sharing BO<sub>6</sub> octahedra form the three-dimensional lattice with every B<sub>8</sub> cube containing one A cation. An ideal cubic perovskite, in which the B-O-B bonds are linear with A located at the center of each B<sub>8</sub> cube (thus forming a AO<sub>12</sub> polyhedron), is expected when the tolerance factor  $\tau$  =  $(r_{\rm O} + r_{\rm A})/\sqrt{2}(r_{\rm O} + r_{\rm B})$  is unity  $(r_{\rm A}$  and  $r_{\rm B}$  are the ionic radii of the A and B cations, respectively, with  $r_0$  as that of the  $O^{2-}$ anion). In most cases, the A cations are small, so that  $\tau < 1$ and the A-O bonds are too long to maintain the ideal cubic structure. Consequently, the A cation moves away from the center of the B<sub>8</sub> cube, which is accompanied by the bending of the B-O-B bonds and the rotation of the BO6 octahedra to form a lower-coordinate  $AO_n$  polyhedron (n < 12) with short A-O bonds. The distortion of the ideal cubic perovskite toward a non-centrosymmetric FE structure requires another local instability apart from  $\tau < 1$ , namely, the second-order Jahn-Teller (SOJT) instability<sup>[4-6]</sup> of the A-site and/or B-site cation.

As found for PbTiO<sub>3</sub>,<sup>[7]</sup> PbVO<sub>3</sub>,<sup>[8]</sup> and BiCoO<sub>3</sub>,<sup>[9]</sup> moving the A cation toward the center of one B<sub>4</sub> face (Figure 1 a) leads to a tetragonal FE structure with space group *P4mm*. The B cations move away from the approaching A cations such that the coordinate environment of each B cation is more like a BO<sub>5</sub> square pyramid with shortened axial B–O

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[\*\*] The work at NCSU was supported by U.S. DOE under Grant No. DE-FG02-86ER45259 and by the resources of the NERSC Center. The work in China was supported by National Natural Science Foundation of China (50121202, 20533030, 10474087), National Key Basic Research Program under Grant No.2006CB922004, the USTC-HP HPC project, and the SCCAS and Shanghai Supercomputer Center.

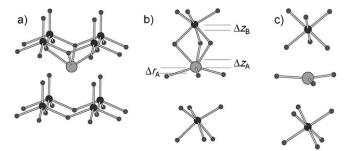


Figure 1. Perspective views of the local structures around the A cation in three perovskites ABO3, where the large, medium, and small circles refer to the A, B, and O atoms, respectively. a) The FE distortion in ABO<sub>3</sub> with space group P4mm. Only eight corner-sharing BO<sub>5</sub> square pyramids and one A cation are shown. b) The FE distortion of ABO3 with space group R3c. For simplicity, only the two BO6 octahedra at the two corners of the B<sub>8</sub> cube along the body-diagonal direction and the  $AO_3$  unit lying between the two octahedra are shown. The  $AO_3$  unit merges with the upper BO<sub>6</sub> octahedron to form a distorted AO<sub>6</sub> octahedron, which shares its face with the upper BO<sub>6</sub> octahedron, resulting in the face-sharing octahedral dimer ABO<sub>9</sub>. c) The PE structure of ABO<sub>3</sub> with space group  $R\bar{3}c$ , by which one FE structure is converted into the other FE structure with opposite polarization. The B cation in each  $BO_6$  octahedron is located at the center of the  $O_6$ octahedron, and the  $AO_3$  unit lying at the midpoint between the two octahedra is planar.

bond. In PbTiO $_3$ , PbVO $_3$ , and BiCoO $_3$ , the  $s^2$  A-site cations Pb $^{2+}$  and Bi $^{3+}$  are susceptible to a SOJT distortion, which mixes the empty 6p orbital of A into the filled  $6s^2$  orbital of A hence forming a lone pair on A. [4] In PbTiO $_3$ , the B-site cation Ti $^{4+}$  (d $^0$ ) is also susceptible to a SOJT distortion, which mixes the empty Ti 3d orbitals into the filled 2p orbitals of the oxygen atom of the axial Ti $^{-}$ O bond. [5,6] Thus, in PbTiO $_3$ , the need to make short Pb $^{-}$ O bonds and the SOJT instabilities of both Pb $^{2+}$  and Ti $^{4+}$  ions cooperate to give rise to the observed FE distortion. [10,11]

Another type of the A-cation displacement from the center of the  $B_8$  cube is the movement toward one corner of the  $B_8$  cube along the body-diagonal direction (that is, the  $C_3$  rotational axis; Figure 1 b) to form a non-centrosymmetric FE trigonal structure with space group R3c, as found for BiFeO<sub>3</sub>.<sup>[12]</sup> The movement of the A cation toward one corner is equivalent to that toward the opposite corner, and the two energetically-equivalent FE structures are mutually interconverted through the centrosymmetric paraelectric (PE) trigonal structure with space group  $R\bar{3}c$  (Figure 1c). <sup>[13,14]</sup> For BiFeO<sub>3</sub>, the FE Curie temperature ( $T_E$  = 1123 K) <sup>[15]</sup> and the three-dimensional antiferromagnetic ordering temperature ( $T_N$  = 650 K) <sup>[16]</sup> are well above room temperature. For

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practical applications, it is highly desirable to find multiferroics with both  $T_E$  and  $T_N$  close to room temperature. Recently, Belik et al.[17] have achieved a partial success toward this goal; they synthesized the indium-based multiferroics  $(In_{1-x}M_x)MO_3$   $(x \approx 0.112-0.176; M = Mn_{0.5}Fe_{0.5})$  that have a  $T_N$  value close to room temperature; however,  $T_E$  is still high (for x = 0.176,  $T_N \approx 270$  K and  $T_E$  is expected to be above 670 K, but the phase is transformed irreversibly to a corundum-type structure above 670 K). These multiferroics, which are isostructural with BiFeO<sub>3</sub>, are quite surprising in two aspects. First, magnetic ordering of a system made up of magnetic ions is generally prevented when disorder is present in the system. However, long-range magnetic ordering occurs near room temperature despite the expected disorder in the A and B sites (that is, the statistical distribution of high-spin Mn<sup>3+</sup> and Fe<sup>3+</sup> cations). Second, it is believed that the strong FE distortion in perovskites with s<sup>2</sup> A-site cations, such as BiFeO<sub>3</sub>, PbTiO<sub>3</sub>, PbVO<sub>3</sub>, and BiCoO<sub>3</sub>, arises from the need to create space for the lone pair on A. However, a strong FE distortion that is almost as strong as found for BiFeO<sub>3</sub> occurs in the indium-based multiferroics, although the A-site cations (In<sup>3+</sup>, Mn<sup>3+</sup>, and Fe<sup>3+</sup>) do not possess a lone pair. Herein, we explore these puzzling observations on the basis of systematic density functional calculations for BiFeO<sub>3</sub> and model perovskites ABO<sub>3</sub> (A = Ga, In, Tl; B = Mn, Fe,  $Mn_{0.5}Fe_{0.5}$ ) that simulate the indium-based multiferroics  $(In_{1-x}M_x)MO_3$ .

Unless stated otherwise, our calculations were carried out for the antiferromagnetic (AFM) states of ABO<sub>3</sub> (A = Ga, In, T1; B = Mn, Fe,  $Mn_{0.5}Fe_{0.5}$ ) and  $BiFeO_3$  in which all nearestneighbor (NN) spins of the B-site cations are antiferromagnetically coupled. In the hexagonal setting of the trigonal space group R3c, a unit cell of these compounds has six layers of B cations perpendicular to the c axis. To simulate the random arrangement of the  $Mn^{3+}$  and  $Fe^{3+}$  ions in A(Mn<sub>0.5</sub>Fe<sub>0.5</sub>)O<sub>3</sub>, we adopted the ordered arrangement in which layers of Mn3+ ions alternate with layers of Fe3+ ions along the c direction. This arrangement is statistically most probable because each B<sub>8</sub> cube is given by Mn<sub>4</sub>Fe<sub>4</sub> such that all NN cations are different. In our GGA + U calculations for the FE structures of ABO<sub>3</sub> (A = Ga, In, Tl; B = Mn, Fe,  $Mn_{0.5}Fe_{0.5}$ ) and BiFeO<sub>3</sub> (with space group R3c), the positions of the atoms were fully optimized but the cell parameters were kept constant at the experimental values. The cell parameters of ABO<sub>3</sub> (A = Ga, In, Tl; B = Mn, Fe,  $Mn_{0.5}Fe_{0.5}$ ) were fixed at those found for  $(In_{1-x}M_x)MO_3$  ( $x \approx 0.176$ ; M =  $Mn_{0.5}Fe_{0.5}$ ).[17]

The results of our structure optimizations for the FE structures of ABO<sub>3</sub> are summarized in terms of the three parameters  $\Delta r_{\rm A}$ ,  $\Delta z_{\rm A}$  and  $\Delta z_{\rm B}$  (Figure 1b), where  $\Delta r_{\rm A}$  is the A-cation displacement from the center of the B<sub>8</sub> cube to the actual position of A in the B<sub>8</sub> cube; the strength of the AO<sub>6</sub> (BO<sub>6</sub>) octahedron distortion is measured by the displacement  $\Delta z_{\rm A}$  ( $\Delta z_{\rm B}$ ) from the center of the O<sub>6</sub> octahedron to the actual position of the A (B) cation. Given that the z axis taken along the c direction of the hexagonal setting (i.e., the body-diagonal direction of the B<sub>8</sub> cube; Figure 1b) shows that  $\Delta z_{\rm A} < 0$  and  $\Delta z_{\rm B} > 0$ . The values of  $\Delta r_{\rm A}$  are summarized in Table 1, and the  $\Delta z_{\rm A}$  and  $\Delta z_{\rm B}$  values are given in Table 2. The  $\Delta r_{\rm A}$  values are substantial, and are largely determined by the

**Table 1:** Values of  $\Delta r_{\Delta}$  calculated for ABO<sub>3</sub>. [a]

Α	$A(Mn_{0.5}Fe_{0.5})O_3$	$AMnO_3$	AFeO <sub>3</sub>	
Ga	0.588	0.550	0.593	
In	0.504	0.479	0.559	
Tl	0.321	0.309	0.339	

[a] B = Mn<sub>0.5</sub>Fe<sub>0.5</sub>, Mn, Fe.  $\Delta r_A$  = 0.295 Å for BiFeO<sub>3</sub>.

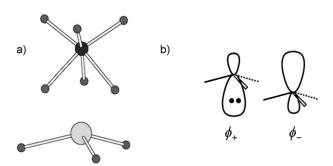
**Table 2:** Values of  $\Delta z_A$  and  $\Delta z_B$  calculated for ABO<sub>3</sub>. [a]

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В	Parameter	GaBO <sub>3</sub>	InBO <sub>3</sub>	TIBO <sub>3</sub>
Mn <sub>0.5</sub> Fe <sub>0.5</sub>	$\Delta z_{\rm B}({\sf Mn})$	0.235	0.185	0.108
	$\Delta r_{\rm A}({\rm Mn})$	-0.360	-0.479	-0.747
	$\Delta z_{\rm B}({\rm Fe})$	0.282	0.234	0.137
	$\Delta r_{\rm A}({\rm Fe})$	-0.318	-0.449	-0.720
Mn	$\Delta z_{B}$	0.226	0.182	0.136
	$\Delta r_{A}$	-0.345	-0.467	-0.745
Fe	$\Delta z_{B}$	0.269	0.216	0.122
	$\Delta r_{A}$	-0.344	-0.468	-0.732

[a] A=Ga, In, Tl; B= Mn<sub>0.5</sub>Fe<sub>0.5</sub>, Mn, Fe.  $\Delta z_A$ =-0.641 Å and  $\Delta z_B$ = 0.193 Å for BiFeO<sub>3</sub>.

A cation. For a given ABO<sub>3</sub> series and BiFeO<sub>3</sub>, the  $\Delta r_A$  values decrease in the order Ga<sup>3+</sup> > In<sup>3+</sup> > Tl<sup>3+</sup> > Bi<sup>3+</sup>, which reflects that the A-cation size increases in the order Ga<sup>3+</sup> < In<sup>3+</sup> < Tl<sup>3+</sup> < Bi<sup>3+</sup>. Table 2 reveals that the  $\Delta z_A$  and  $\Delta z_B$  values are substantial and are largely determined by the A cation; the  $|\Delta z_A|$  values increase in the order Ga<sup>3+</sup> < In<sup>3+</sup> < Tl<sup>3+</sup>, and the  $\Delta z_B$  values decrease with increasing  $|\Delta z_A|$ . The  $\Delta z_A$  and  $\Delta z_B$  values of BiFeO<sub>3</sub> are close to those of TlFeO<sub>3</sub>.

As described above, the FE distortion of the indium-based multiferroics ( $In_{1-x}M_x$ )MO<sub>3</sub> ( $x\approx0.112$ –0.176; M = Mn<sub>0.5</sub>Fe<sub>0.5</sub>) is very similar in nature and magnitude to that found in BiFeO<sub>3</sub>, which suggests that the origin of their FE distortions is identical. The face-sharing octahedral dimer ABO<sub>9</sub> (Figure 1b) along the body-diagonal direction of the B<sub>8</sub> cube can be regarded as a result of the interaction between the BO<sub>6</sub> octahedron and the AO<sub>3</sub> pyramid (Figure 2a). As depicted in Figure 2b, the AO<sub>3</sub> pyramid has two hydrid orbitals  $\phi_+$  and  $\phi_-$  made up of the ns and np orbitals (n=4, 5, 6 for Ga<sup>3+</sup>,  $In^{3+}$ , and  $II^{3+}$ , respectively). <sup>[4]</sup> In the case of the s<sup>2</sup> cation Bi<sup>3+</sup>, the  $\phi_+$  orbital is doubly occupied to become a lone pair, whilst the  $\phi_-$  orbital is empty. In the case of the s<sup>0</sup> cations Ga<sup>3+</sup>,  $In^{3+}$  and



**Figure 2.** a) The face-sharing octahedral dimer ABO<sub>9</sub> in the FE structure of ABO<sub>3</sub> with space group R3c in terms of the BO<sub>6</sub> octahedron and the AO<sub>3</sub> pyramid. b) The two hybrid orbitals  $\phi_+$  and  $\phi_-$  of the AO<sub>3</sub> pyramid.

 $Tl^{3+}$ , both  $\phi_{+}$  and  $\phi_{-}$  orbitals are empty. The energy lowering associated with the interaction of the AO<sub>3</sub> pyramid with the BO<sub>6</sub> octahedron occurs primarily from the interaction of the empty  $\phi_{-}$  orbital with the filled 2p orbitals of the three oxygen atoms that form the A-O bonds. This interaction is the driving force for the SOJT distortion of the A-site cations, and is therefore independent of whether the A-site cation is an s<sup>0</sup> or an s<sup>2</sup> cation. It was shown above that the  $|\Delta z_A|$  values increase in the order  $Ga^{3+} < In^{3+} < Tl^{3+}$ , whereas the  $\Delta z_B$ values decrease with increasing  $|\Delta z_A|$ . The extent of the A<sup>3+</sup> cation displacement in the AO<sub>6</sub> octahedron from the center of the O<sub>6</sub> octahedron is related to the extension of the  $A^{3+}$ -cation np orbital (n = 4, 5, and 6 for  $Ga^{3+}$ ,  $In^{3+}$ , and  $Tl^{3+}$ , respectively); the maximum overlap of the  $A^{3+}$  cation  $\phi_{-}$ orbital (hence the np orbital) with the O 2p orbitals of the three oxygen atoms forming the A-O bonds should occur at a larger  $|\Delta z_A|$  value as the A cation is varied from  $Ga^{3+}$  to  $In^{3+}$ to Tl<sup>3+</sup>. The involvement of the In<sup>3+</sup> cation 5p orbitals in the FE distortion of the perovskite  $InBO_3$  (B=Mn, Fe, Mn<sub>0.5</sub>Fen<sub>0.5</sub>) can be seen from the plots of the density of states (DOS) calculated for the FE and PE structures of InFeO<sub>3</sub> (Figure 3). The PE structure has more unoccupied

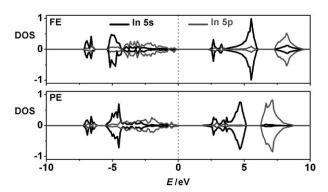


Figure 3. Plots of the projected DOS (states/eV per two formula units) calculated for the In 5s and In 5p orbitals in the FE and PE structures of InFeO<sub>3</sub>. The up- and down-spin densities are indicated by plus and minus DOS values, respectively.

In 5p states in a lower energy region than does the FE structure. This result means that more In 5p states are involved in bonding with the O 2p states in the FE state than in the PE state, as expected from the SOJT distortion involving the In<sup>3+</sup> cation. Similar interactions should occur between the np orbitals of the A-site cation and the O 2p orbitals in the FE structure of ABO<sub>3</sub> with space group *P4mm*, as found for PbVO<sub>3</sub>.<sup>[18]</sup>

We then examined the FE polarization of  $(In_{1-x}M_x)MO_3$   $(x \approx 0.176; M = Mn_{0.5}Fe_{0.5})$  by considering only the perovskites AFeO<sub>3</sub> (A = Ga, In, Tl), because the FE distortion is largely determined by the A cation, as discussed above. As found for BiFeO<sub>3</sub>,  $^{[13,14]}$  the most probable PE structure of AFeO<sub>3</sub> (A = Ga, In, Tl) has the centrosymmetric structure with space group  $R\bar{3}c.^{[19]}$  The rhombohedral setting of this structure has the positions Fe(0,0,0),  $In(^1/_4,^1/_4,^1/_4)$ , and  $O(x,^1/_2-x,^1/_4)$ . Our structure optimizations for the PE structures lead to x = 0.6023, 0.6399, 0.6405, and 0.6707 for

GaFeO<sub>3</sub>, InFeO<sub>3</sub>, TIFeO<sub>3</sub>, and BiFeO<sub>3</sub>, respectively. Given the FE and PE structures of AFeO<sub>3</sub>, we calculated their FE polarizations P (along the body-diagonal direction of the B<sub>8</sub> cube) on the basis of the Berry phase method, [20] which leads to 46, 74, and 139 μCcm<sup>-2</sup> for GaFeO<sub>3</sub>, InFeO<sub>3</sub>, and TIFeO<sub>3</sub>, respectively. These values are comparable to 98.7 μCcm<sup>-2</sup> calculated for BiFeO<sub>3</sub>. [13,14] Experimentally, the FE polarization of (In<sub>1-x</sub>M<sub>x</sub>)MO<sub>3</sub>[17] is comparable in magnitude to that of BiFeO<sub>3</sub>[13,14] ( $P \approx 1 \, \mu \text{Ccm}^{-2}$  for x = 0.143 versus 2–6 μC cm<sup>-2</sup>).

To understand why  $(In_{1-x}M_x)MO_3$   $(M = Mn_{0.5}Fe_{0.5})$  has a significantly lower  $T_N$  (about 270 K for  $x \approx 0.176$ ) than does BiFeO<sub>3</sub> (650 K), it is necessary to evaluate the NN spin exchange parameters J of InMO<sub>3</sub> (M = Mn, Fe, Mn<sub>0.5</sub>Fe<sub>0.5</sub>) and BiFeO<sub>3</sub> by calculating the energies of the two ordered spin states for InMO3 and BiFeO3; that is, the AFM state considered so far and the ferromagnetic (FM) state in which all NN spins are ferromagnetically coupled. By mapping the energy difference between the two states onto the corresponding energy difference expected from the spin Hamiltonian expressed in terms of  $J_{s}^{[21,22]}$  we obtain J = -18.8, -14.6, +10.4, and -6.1 meV for BiFeO<sub>3</sub>, InFeO<sub>3</sub>, InMnO<sub>3</sub>, and In(Mn<sub>0.5</sub>Fe<sub>0.5</sub>)O<sub>3</sub>, respectively. Namely, the AFM state is more stable than the FM state for InFeO<sub>3</sub> and In(Mn<sub>0.5</sub>Fe<sub>0.5</sub>)O<sub>3</sub>, as found for BiFeO<sub>3</sub>, whereas the opposite is the case for InMnO<sub>3</sub>. [23,24] Note that the J value of In(Mn<sub>0.5</sub>Fe<sub>0.5</sub>)O<sub>3</sub> is smaller than those of InFeO<sub>3</sub> and BiFeO<sub>3</sub> by a factor of about 2.5–3. In the mean field approximation, [25]  $T_N$  scales linearly with |J|, so that the  $T_N$  of  $In(Mn_{0.5}Fe_{0.5})O_3$ is estimated to be circa 220-260 K from  $T_N = 650 \text{ K}$  for BiFeO<sub>3</sub>. The latter value is quite close to that found for the indium-based multiferroics  $(In_{1-x}M_x)MO_3$   $(M = Mn_{0.5}Fe_{0.5})$ (i.e., about 270 K for  $x \approx 0.176$ ).<sup>[17]</sup> Therefore, the threedimensional magnetic ordering of  $(In_{1-x}M_x)MO_3$  (M =Mn<sub>0.5</sub>Fe<sub>0.5</sub>) near room temperature arises from two factors: one is that the statistically most probable distribution of the  $Mn^{3+}$  and  $Fe^{3+}$  ions in the b sites leads to the three-dimensional arrangement in which the spin exchange paths between these magnetic ions are largely given by the Mn-O-Fe superexchanges: the other factor is that the Mn-O-Fe superexchange is weaker than the Fe-O-Fe superexchange by a factor of about three.

Finally, we addressed the question of how to lower  $T_E$  of perovskite multiferroics with s<sup>0</sup> A-site cations by considering AFeO<sub>3</sub> (A = Ga, In, Tl, Bi). It is not surprising that the indium-based multiferroics  $(In_{1-x}M_x)MO_3$   $(M = Mn_{0.5}Fe_{0.5})$ have a high  $T_E$  and a high FE polarization because their FE distortion is nearly as strong as that found for BiFeO<sub>3</sub>. Qualitatively,  $T_E$  of AFeO<sub>3</sub> (A = Ga, In, Tl, Bi) is expected to be lowered when the stability of the FE structure with respect to the PE structure is reduced. Our calculations for the optimized FE and PE structures of AFeO<sub>3</sub> predict that the FE structure is more stable than the PE structure by 0.31, 0.45, 0.20, and 0.32 eV per formula unit for A = Ga, In, Tl, and Bi, respectively, suggesting that the tendency for the A-site cation to undergo an FE distortion should decrease in the order  $In^{3+} > Bi^{3+}$ ,  $Ga^{3+} > Tl^{3+}$ . Therefore,  $TlFeO_3$  would have a lower  $T_E$  value than does BiFeO<sub>3</sub>. It would be of interest to prepare the thallium analogue of the indium-based multi-

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ferroics, such as  $(Tl_{1-x}M_x)MO_3$   $(M=Mn_{0.5}Fe_{0.5})$ . Another interesting compound would be  $(Ga_{1-x}M_x)MO_3$   $(M=Mn_{0.5}Fe_{0.5})$ , because the presence of  $Mn^{3+}$  and  $Fe^{3+}$  ions at the A-cation sites would reduce the tendency for the  $Ga^{3+}$  ions to undergo an FE distortion, thus lowering  $T_E$ .

In summary, our work explains why  $T_N$  of the indiumbased multiferroic perovskites  $(In_{1-x}M_x)MO_3$   $(M=Mn_{0.5}Fe_{0.5})$  is near room temperature and why the perovskites with  $s^0$  and  $s^2$  A-site cations have similar ferroelectric distortions. Possible perovskites with  $T_E$  lower than that of  $(In_{1-x}M_x)MO_3$  are expected from  $(Tl_{1-x}M_x)MO_3$   $(M=Mn_{0.5}Fe_{0.5})$  and  $(Ga_{1-x}M_x)MO_3$   $(M=Mn_{0.5}Fe_{0.5})$ .

## **Experimental Section**

The calculations employed the projector-augmented wave method  $^{[26]}$  encoded in the Vienna ab initio simulation package,  $^{[27]}$  the PW91 functional form  $^{[28]}$  of the generalized gradient approximation (GGA), and the plane-wave cut-off energy of 400 eV. The GGA plus on-site repulsion U method  $^{[29]}$  (GGA + U) was used to properly describe the electron correlation associated with the Mn and Fe 3d states with several values of U for both manganese and iron. Similar results are obtained from the use of other U values, so we report only those from U=4.5 eV.

Received: October 24, 2009 Revised: December 17, 2009 Published online: February 1, 2010

**Keywords:** density functional calculations · ferroelectric distortion · indium · Jahn–Teller distortion · perovskite phases

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